

Stereoselective ($2\pi + 2\pi$) Photocycloaddition of Arylalkenes to Chrysene

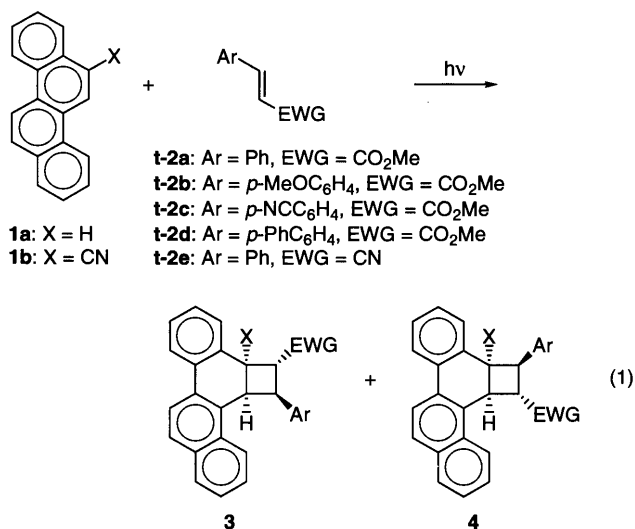
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Photocycloaddition of electron-deficient arylalkenes to chrysene stereoselectively occurred at 5,6-position of chrysene ring to give two ($2\pi + 2\pi$) photocycloadducts through sandwich-type singlet exciplexes. Photocycloreversion took place via the excited singlet states of the photocycloadducts.

Photocycloaddition of unsaturated molecules to aromatic rings is not only an interesting subject on organic photochemistry, but a useful synthetic method for the synthesis of conformationally rigid cyclobutanes.¹ Most of studies were developed through the inter- or intramolecular photocycloaddition of unsaturated compounds to benzene, naphthalene, anthracene, phenanthrene, and pyrene rings.² On the contrary, photocycloaddition to more condensed polynuclear aromatic hydrocarbons is scarcely known.³ Chrysene is one of the most usable aromatic hydrocarbons as photosensitizer,⁴ but the study about the photocycloaddition to chrysene ring is hitherto only known about the ($4\pi + 2\pi$) photocycloaddition with maleic anhydride to give a benzopyrene derivative.⁵ Here we describe a novel ($2\pi + 2\pi$) photocycloaddition of arylalkenes to chrysene ring.



Irradiation of a benzene solution containing chrysene (**1a**) and methyl *trans*-cinnamate (**t-2a**) by a high-pressure Hg lamp through Pyrex filter under argon atmosphere gave two regioisomeric ($2\pi + 2\pi$) photocycloadducts (**3aa** and **4aa**) in a stereoselective manner (eq 1, Table 1). In this photoreaction, the cycloadduct derived from *cis*-alkene was not detected at all, although *cis*-isomer (**c-2a**) was produced in detectable amounts. The yields of products increased with increasing the irradiation time less than 2 h (runs 1–3). When CH₃CN, THF, and CH₂Cl₂ were used as a solvent, the efficiency was lower than that in benzene (runs 4–6). Methoxy group on *para* position of methyl

trans-cinnamate accelerated the reaction, however, introduction of *p*-cyano group caused the almost recovery of **1a** (runs 7, 8). Reaction with *p*-phenyl derivative (**t-2d**) gave poor selectivity of **3** / **4** in moderate yields of products (runs 9, 10). Reaction of **1a** with cinnamitrile (**t-2e**) also proceeded (run 11). Photocycloaddition of **t-2a** to 6-cyanochrysene (**1b**) occurred at 5,6-position of **1b** in low efficiency compared with **1a** (runs 12, 13), but the photoreaction of **1b** with **t-2b** proceeded in >10 : 1 selectivity (run 14). The other alkenes such as anethole, *trans*-stilbene, *trans*-crotonitrile, α -cyanostilbene, and benzylidenemalonitrile did not add to **1a** under the same irradiation conditions.⁶

Table 1. ($2\pi+2\pi$) Photocycloaddition of chrysene (**1**) with electron-deficient arylalkenes (**2**)^a

Run	1	2	Solvent	Irradn Time / h	Yields ^b / %		Recovery of 1 / %
					3	4	
1	1a	t-2a	benzene	0.5	26	5	69
2	1a	t-2a	benzene	2	46	10	35
3	1a	t-2a	benzene	5	44	9	27
4	1a	t-2a	CH ₃ CN	2	25	7	45
5	1a	t-2a	THF	2	13	3	66
6	1a	t-2a	CH ₂ Cl ₂	2	<1	<1	83
7	1a	t-2b	benzene	2	60	13	16
8	1a	t-2c ^c	benzene	2	<1	<1	>99
9	1a	t-2d ^c	benzene	2	24	10	66
10	1a	t-2d ^c	benzene	5	24	14	61
11	1a	t-2e	benzene	2	23	6	69
12	1b	t-2a	benzene	2	4	<1	71
13	1b	t-2a	benzene	5	7	<1	59
14	1b	t-2b	benzene	2	53	5	24

^a[**1**] = 0.01 M, [**2**] = 0.1 M. ^bYields were determined by ¹H NMR using *o*-xylene as an internal standard. ^c[**t-2c-d**] = 0.02 M.

The structures of the products were determined by their spectral and analytical data, especially by ¹H NMR. The structure of **3aa** was confirmed by X-ray analysis shown in Figure 1.

The fluorescence of **1a** in benzene was efficiently quenched by **t-2a** ($k_q = 1.21 \times 10^9 \text{ M}^{-1}\text{s}^{-1}$). The formation of photocycloadducts was not sensitized by the addition of benzophenone, and was not quenched by molecular dioxygen. From these facts, the regio- and stereoselectivity to give **3** and **4** was explained by two possible structures of sandwich-type singlet exciplexes formed via excited singlet state of **1a** shown in Scheme 1. Aromatic groups of the electron-deficient arylalkenes might favor to overlap on the naphthalene ring than on the benzene ring. Higher efficiency in benzene than other solvents and low selectivity in the reaction with **t-2d** were also rationalized by this mechanism. Arylalkene having electron-donating substituent gave photocycloadducts efficiently. Higher regioselectivity of the reaction of **1b** with **t-2b** may be explained by their dipole moments in the excited states.^{1c}

UV-vis absorption maximum of **3aa** showed at 334 nm.⁷

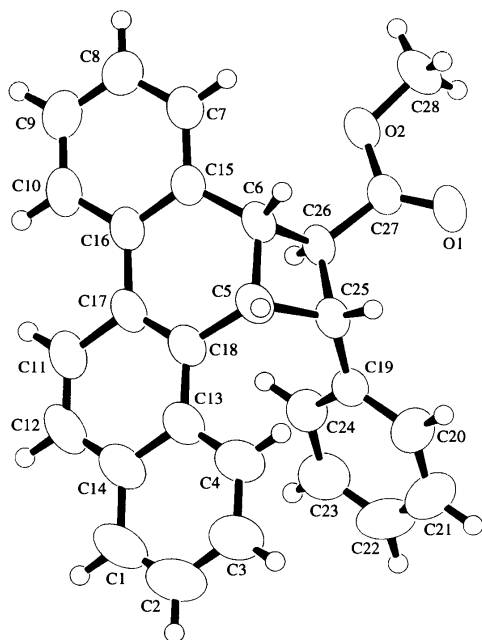
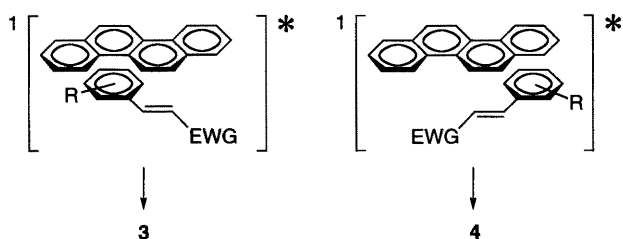
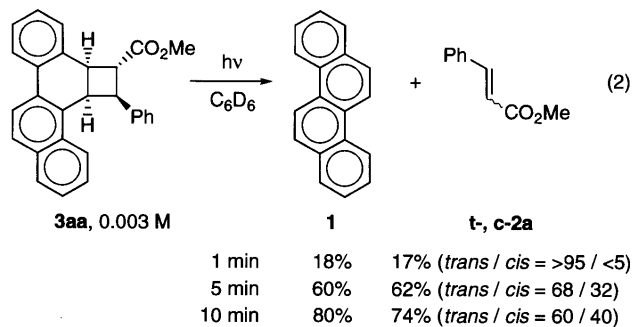


Figure 1. ORTEP drawing of **3aa**. $C_{28}H_{22}O_2$, FW = 390.48, monoclinic, $C2/c$ (#15), $Z = 8$, $R1 = 0.053$, $a = 17.833(4)$, $b = 10.012(4)$, $c = 23.375(5)$ Å, $\beta = 103.03(2)^\circ$, $V = 4066(1)$ Å³, $D_{\text{calcd}} = 1.276$ g cm⁻³. Selected bond lengths (Å); C(5)-C(6): 1.541(3), C(6)-C(26): 1.554(3), C(26)-C(25): 1.540(3), C(25)-C(5): 1.582(3).



Scheme 1. Structures of two possible exciplexes to give photocycloadducts.

Therefore, photocycloreversion from **3** and **4** to **1** and **2** competitively occurred in this reaction.⁸ Photoirradiation of isolated **3aa** in benzene-*d*₆ was monitored by ¹H NMR spectroscopy (eq 2). At the initial stage of this cycloreversion, **2a** was obtained only its *trans*-isomer. The prolonged irradiation caused the isomerization of *t*-**2a** to *c*-**2a**. These results support



the fact that the photocycloreversion occurs via excited singlet state of **3aa**.

In conclusion, we have found the first example of $(2\pi + 2\pi)$ photocycloaddition of chrysenes with alkenes. The highly regioselective photocycloaddition of *t*-**2b** to **1b** has been achieved by the combination of π - π overlapping and favorable dipole moment in the exciplex state. The scope and mechanism in detail are now under investigation.

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- Prolonged irradiation of **1a** with α -cyanostilbene in benzene afforded $(2\pi + 2\pi)$ photocycloadduct: R. A. Caldwell, personal communication.
- 1a**: $\lambda_{\text{max}} = 320$ nm ($\epsilon = 1.8 \times 10^4$), **3aa**: $\lambda_{\text{max}} = 334$ nm ($\epsilon = 8.1 \times 10^3$), both in cyclohexane.
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